

FUSHE2012

Erbismühle - Weilrod, Germany MAY 13th-16th 2012

ENSAR-ECOS Workshop on Future SuperHeavy Element Strategy

Summary Experiment

R.-D. Herzberg

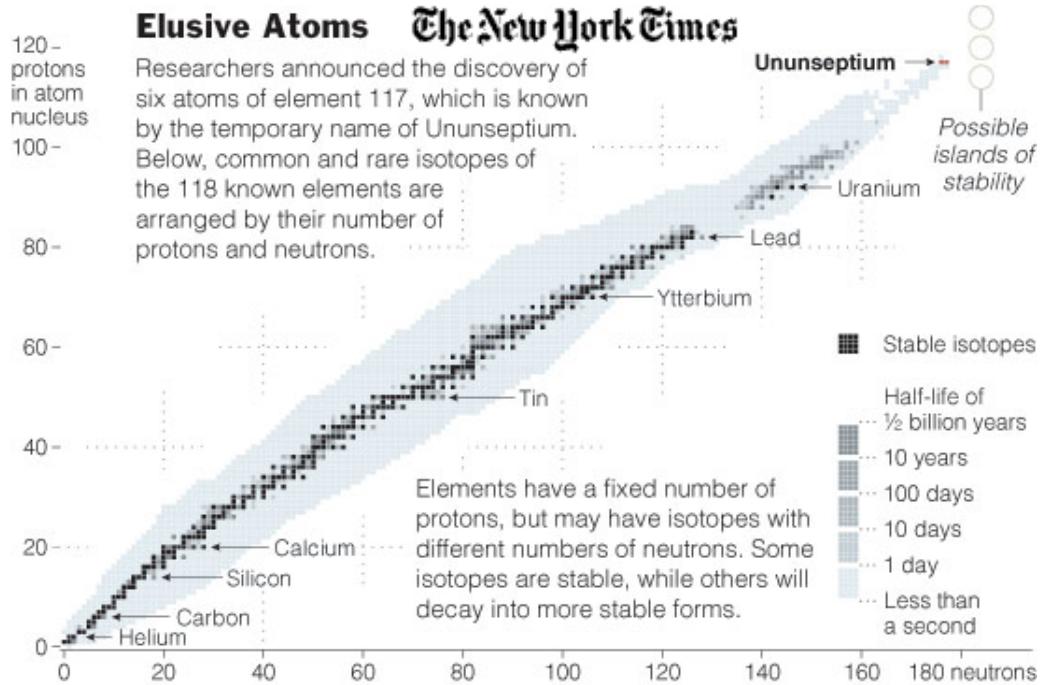
Ch.E. Düllmann

Discovery of a new element is as fascinating to the public as landing on the moon (or Mars)

M. Stoyer

"The question we're trying to answer is, 'Does the periodic table come to an end, and if so, where does it end?' "

Kenton Moody -- LLNL



↑
Yu. Oganessian

Electronic structure
of SHE-atoms

Atomic Physics

Chemical properties of the SHE

Chemistry

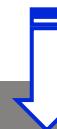
Nuclear structure and decay
properties of the SHN

Nuclear Physics

Nuclear theory

Search for
new shells

$A=294$



Search for SHE
in Nature

Astrophysics

$A=252$

No

SHE

-7

-5

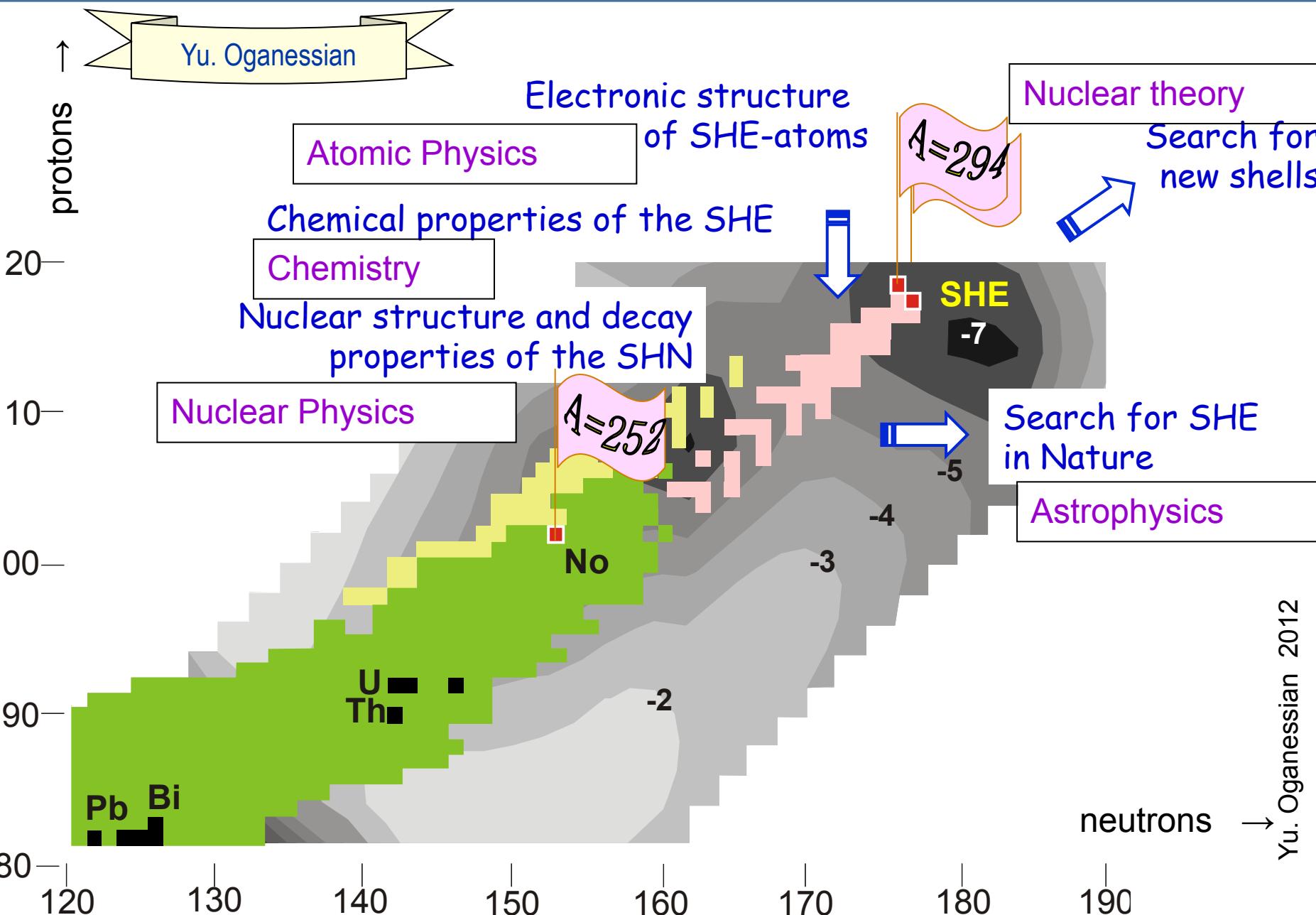
-4

-3

-2

neutrons

→ Yu. Oganessian 2012

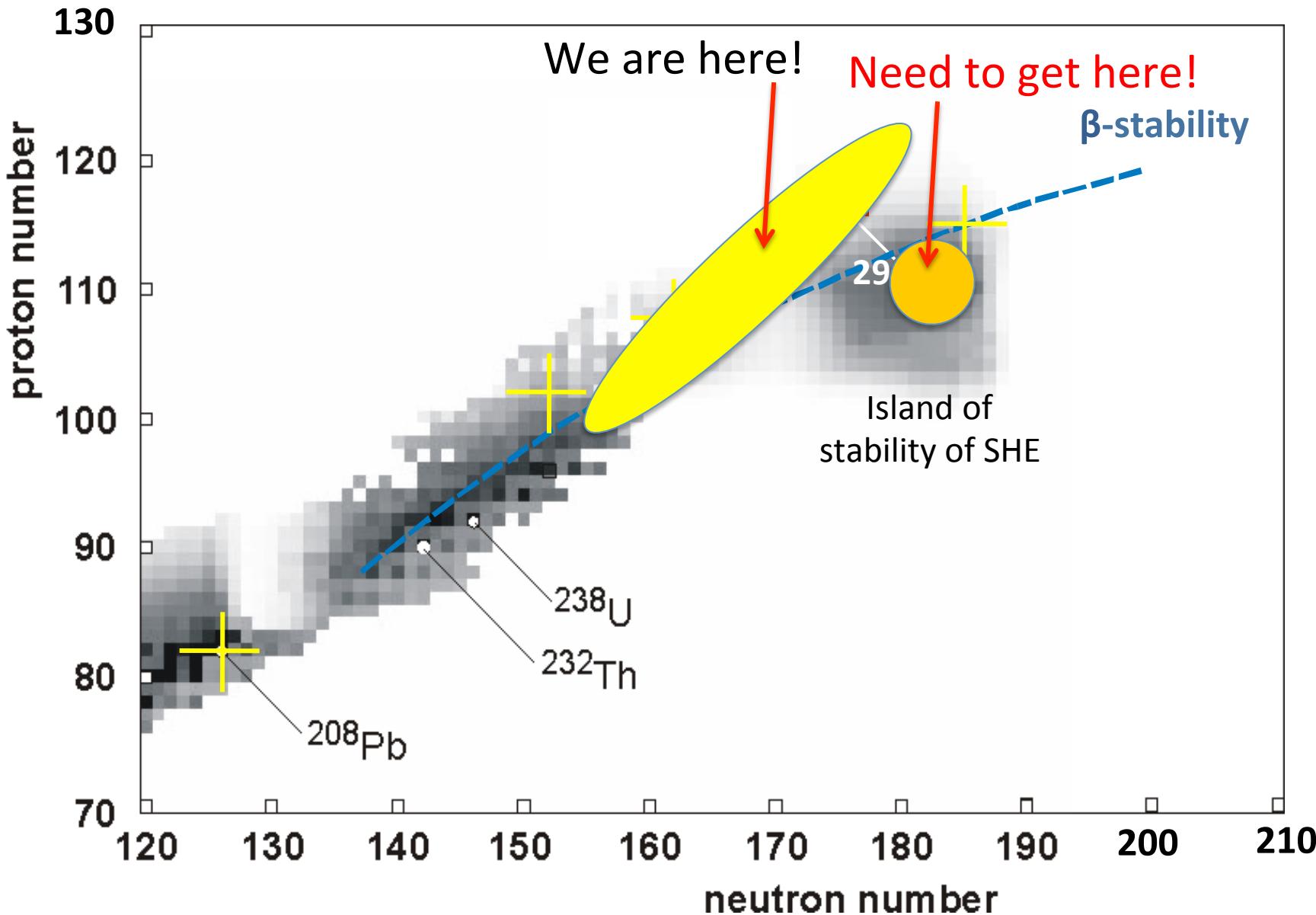


Scientific Questions

- What is the heaviest element we can make?
- What is the heaviest nucleus we can make and what are the best methods to do so?
- What is the nuclear structure in the heaviest nuclei?
- What are the chemical and atomic properties of the heaviest elements and is chemical periodicity altered as relativistic effects become more important?
- Are there astrophysical scenarios in which SHEs are produced or does fission limit the production of SHEs? Do the heaviest elements exist in nature?
- (How) Can we produce weighable quantities on earth?

What is the heaviest element?

- Where is the island of stability?
- What are the best methods to create them?
- What determines their survival and stability?
- What are their properties?



Modern Periodic Table of Elements

1
H
2
Li
Be
11
Na
Mg

13	14	15	16	17	18
C	N	O	S	Cl	He

Should we pick up U+U studies again?
In a fixed-target or in a collider experiment?

Ba	La	→	Hf	Ta	74	75	76	77	78	79	Ag	Cd	In	50	51	52	53	54
87	88	89	104	105	106	107	108	109	110	111	Au	Hg	Tl	Sn	Sb	Te	I	Xe
Fr	Ra	Ac	→	Rf	Db	Sq	Bh	Hs	Mt	Ds	Au	Hg	Tl	Pb	Bi	Po	At	Rn
(119)	(120)	(121)									111	112	113	114	115	116	117	118

?

Z=173

Lanthanides →	58	59	60	61	62	63	64	65	66	67	68	69	70	71
	Ce	Pr	Nd	Pm	Sm	Eu	Gd	Tb	Dy	Ho	Er	Tm	Yb	Lu

Actinides →	90	91	92	93	94	95	96	97	98	99	100	101	102	103
	Th	Pa	U	Np	Pu	Am	Cm	Bk	Cf	Es	Fm	Md	No	Lr

Superactinides → (122 - 155)

New Elements / Isotopes

Observed are mainly integral quantities:

- The atom itself!
- Decay mode
- Half-life
- Q-value or TKE

A rich and growing body of data!

Creation

- Reaction mechanisms?
- Best target/beam combinations?
- How reliably is theory predicting cross sections?

1. Theoretical models of formation dynamics (the problems to be solved)

Experimental possibilities

- Fusion reactions (new SH elements and isotopes)
- Transfer reactions (new neutron-rich SH nuclei)
- Neutron capture (SHE in nature)

2. SHE experiments (what could be really done within the next few years)

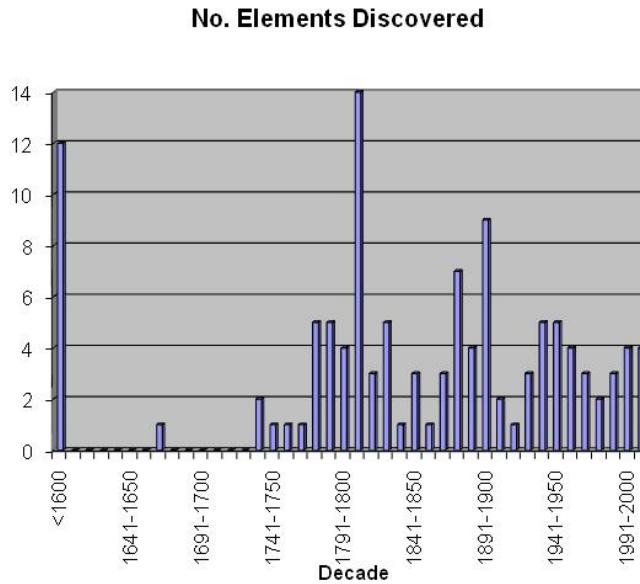
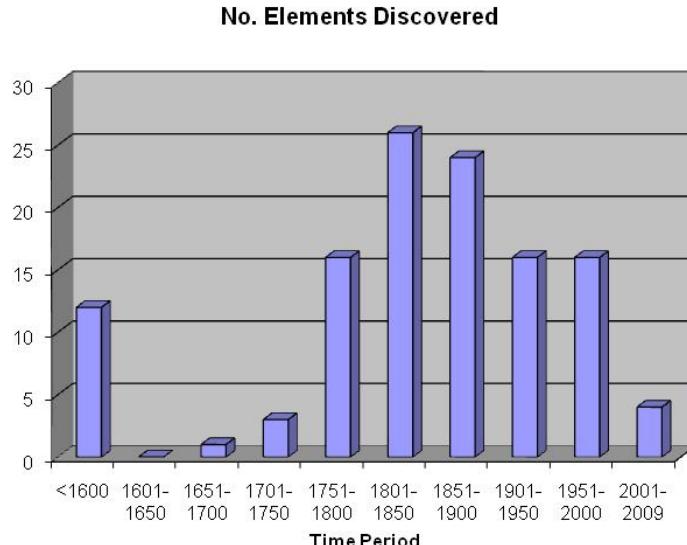
Valeriy Zagrebaev

Flerov Laboratory of Nuclear Reactions, JINR, Dubna

for “Future of Super Heavy Elements”, May 14, 2012, Weiltal, Germany

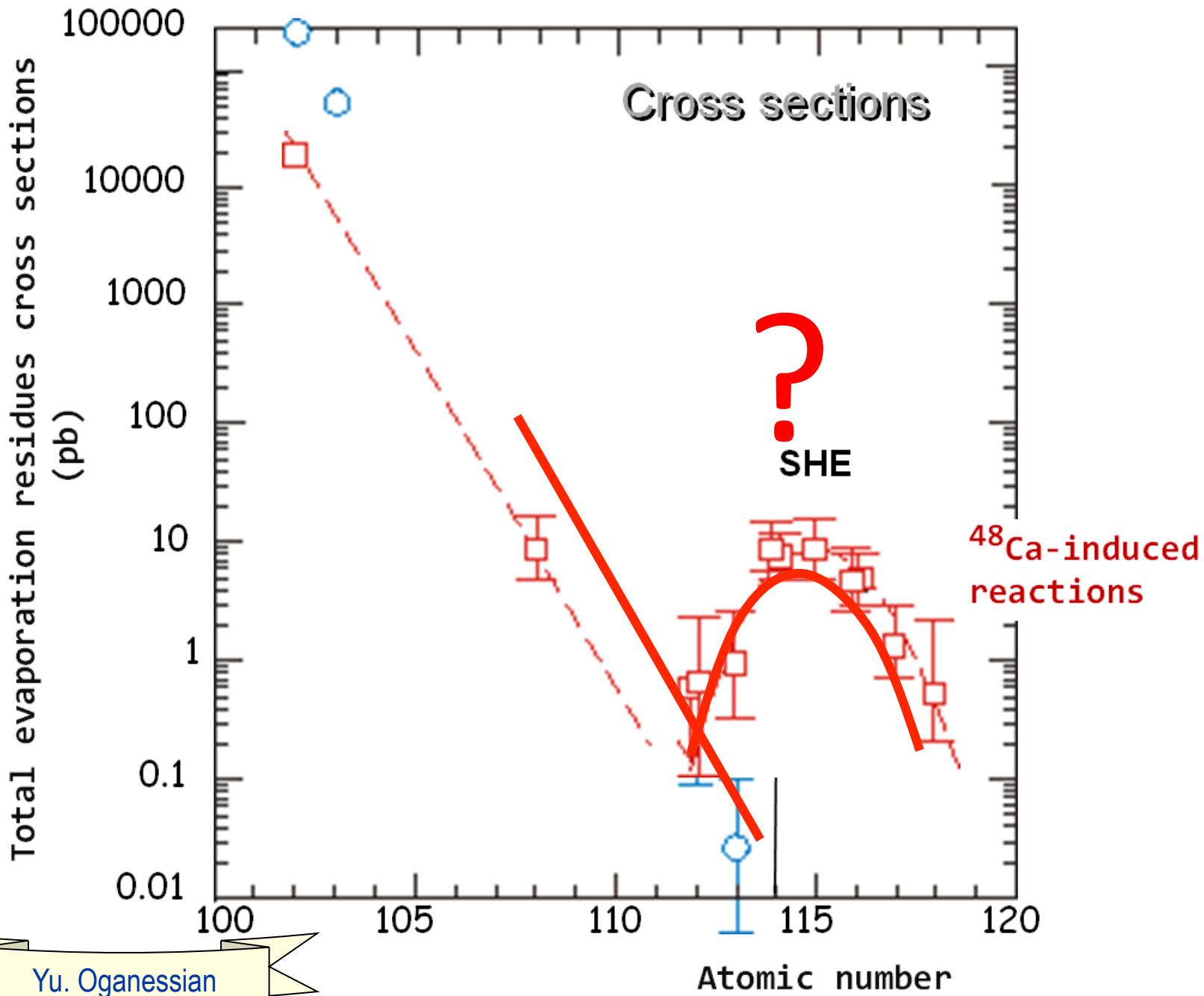
New element discovery has progressed steadily since the 1700's

M. Stoyer



On average, just under 20 elements have been discovered every half-century

On average, about 4 elements have been discovered every decade



Making new elements

E120

Z _{Beam}	Beam	Target	Asymmetry	E* @ B _{Bass}
22	⁵⁰ Ti	²⁴⁹ Cf		31.7
23	⁵¹ V	²⁴⁹ Bk		35.9
24	⁵⁴ Cr	²⁴⁸ Cm		33.0
25	⁵⁵ Mn	²⁴³ Am		34.5
26	⁵⁸ Fe	²⁴⁴ Pu		33.9
27	⁵⁹ Co	²³⁷ Np		32.9
28	⁶⁴ Ni	²³⁸ U		27.3

Similar

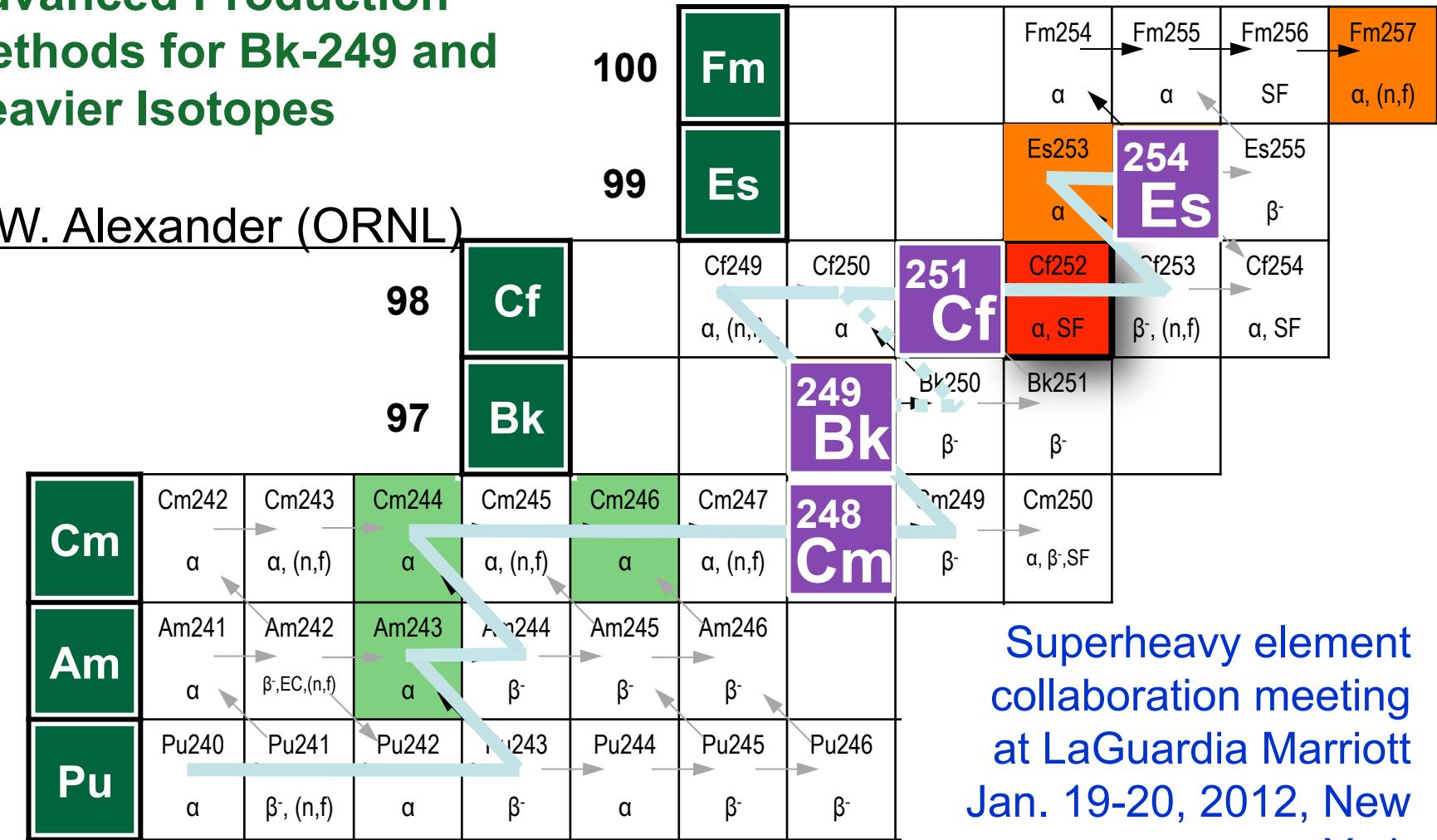
+²⁴⁹Ti

Strategic aspect:
The community depends on
access to transuranium
isotopes in mg quantities!

Transuranium Nuclide Production Paths

Advanced Production Methods for Bk-249 and Heavier Isotopes

C.W. Alexander (ORNL)



Superheavy element
collaboration meeting
at LaGuardia Marriott
Jan. 19-20, 2012, New
York

Yu. Oganessian

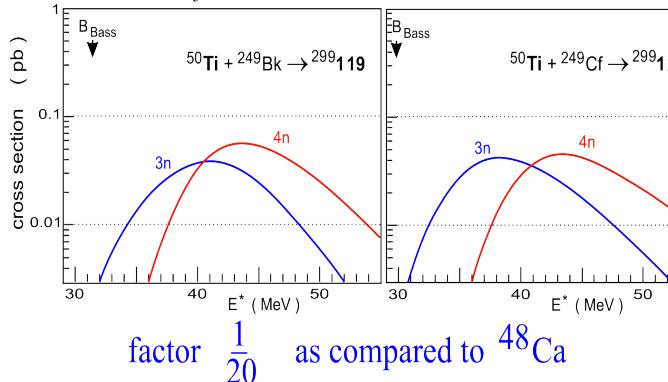
Fusion evaporation reactions

H. Savajols

□ High intense stable beams beyond ^{48}Ca

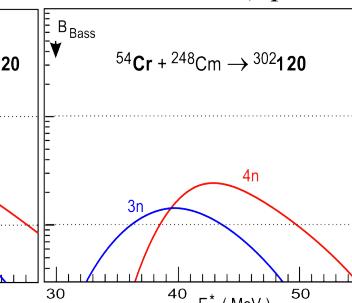
Ti beam:

scheduled for 2012 at TASCA



Cr beam:

started at SHIP (April 23)



V. Zagrebaev et al.

Fusion reactions to $Z=120$



Dubna : $\sigma < 0.4\text{pb}$



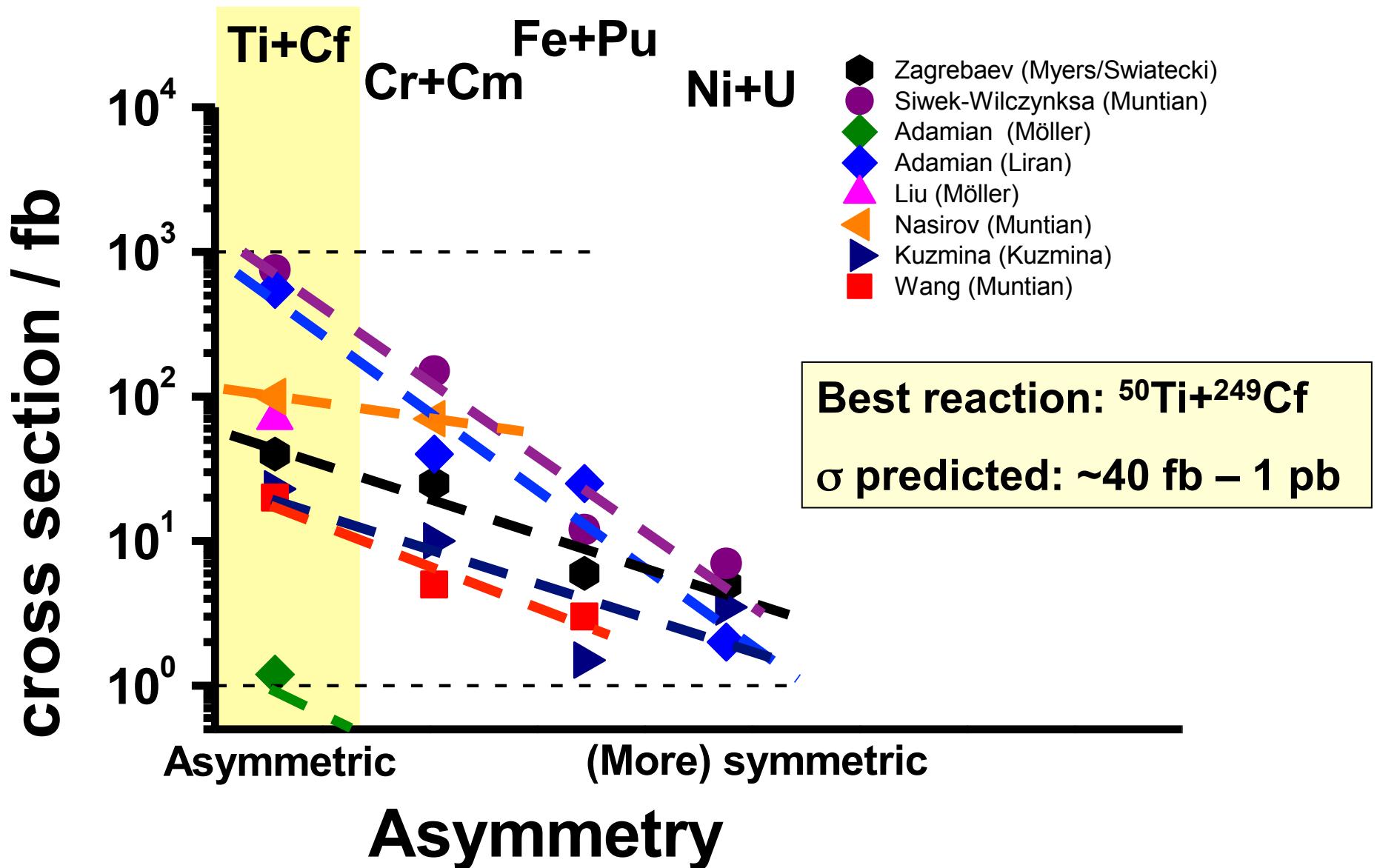
GSI: $\sigma < 0.1\text{ pb}$



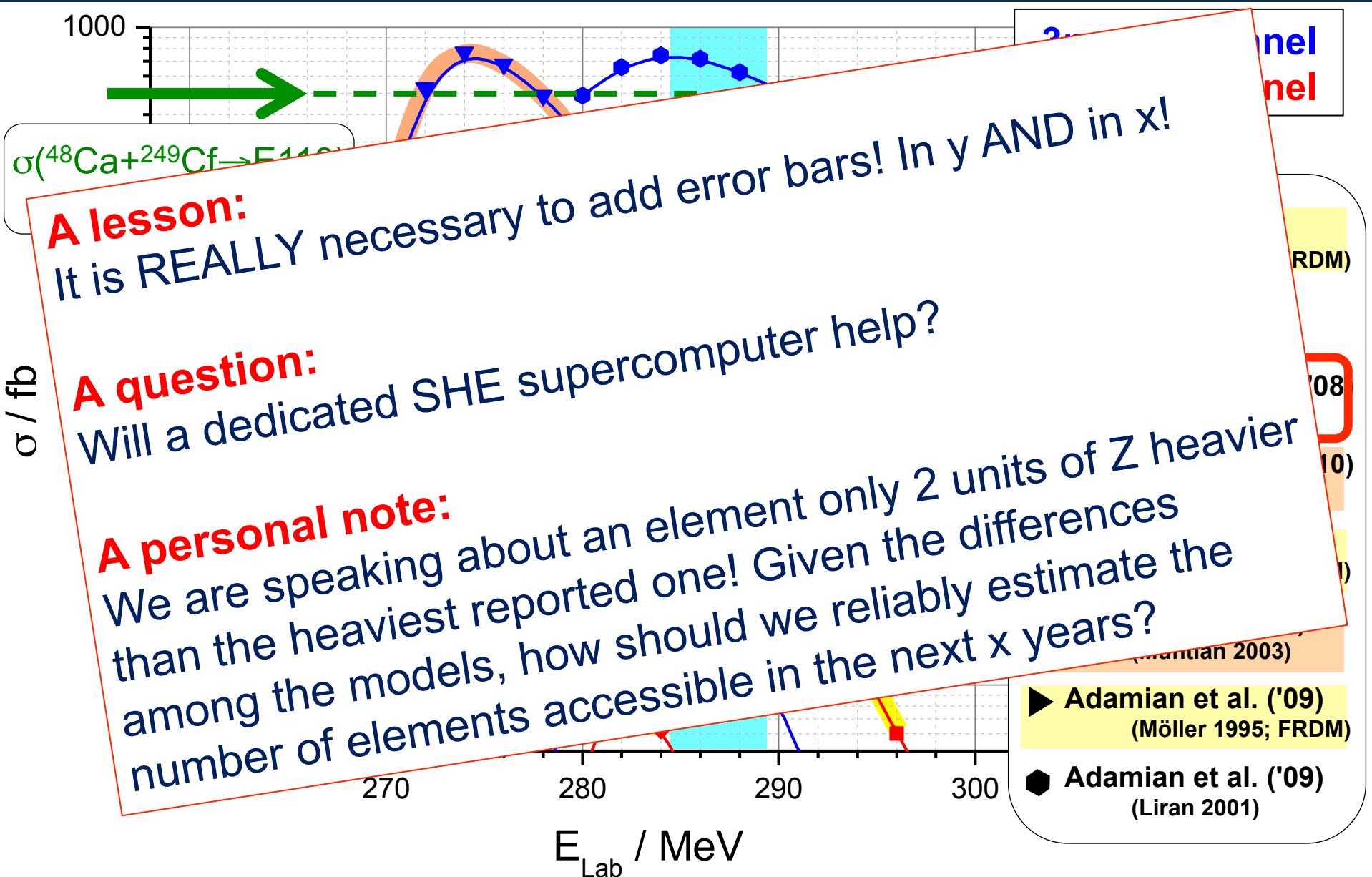
Today: $I \approx 5 \times 10^{12}/\text{s} \rightarrow \sigma_{\text{limit}} \approx 0.1\text{ pb}$

Future: $I \approx 10^{14}/\text{s} \rightarrow \sigma_{\text{limit}} \approx 1 \text{ fb}$

Cross sections: current predictions from theory



$^{50}\text{Ti} + ^{249}\text{Cf}$ Excitation Function

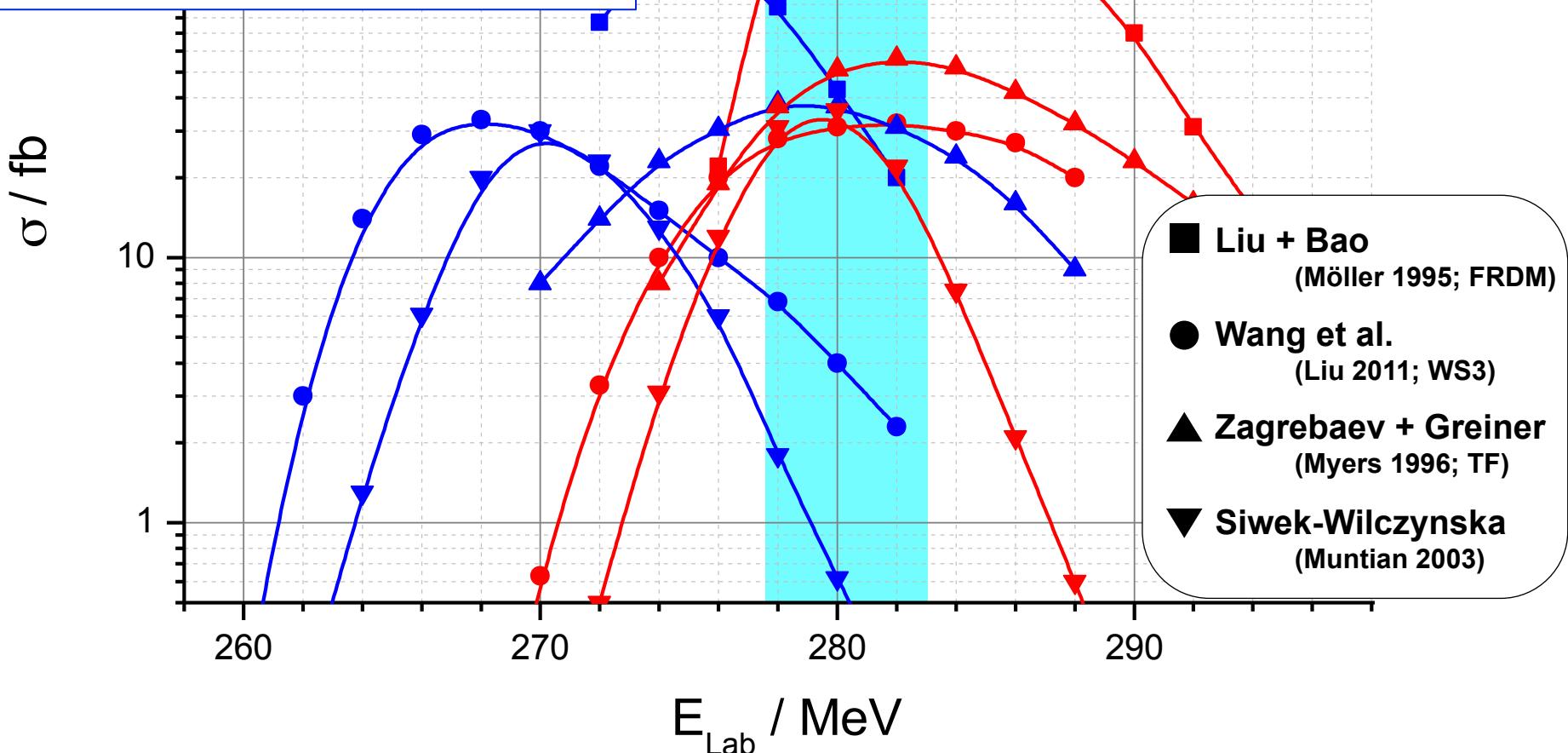


$^{50}\text{Ti} + ^{249}\text{Bk}$ Excitation Function

Agreement 1:
4n is larger than 3n

Agreement 2:
Position (in E) of maximum

3n exit channel
4n exit channel



The **TASCA** Element 119 Collaboration



GSI Darmstadt (D)

D. Ackermann, M. Block, F.P. Heßberger, A. Hübner, E. Jäger,

P. Korten, K. Korten, P. Lederer, P. Pfeiffer

March 6:

Arrival of ^{249}Bk in
Mainz

March 23:

Arrival of Targets at
GSI

April 12:

Mounting Targets in
TASCA

April 14:

Begin Search for
Element 119



U Bogota



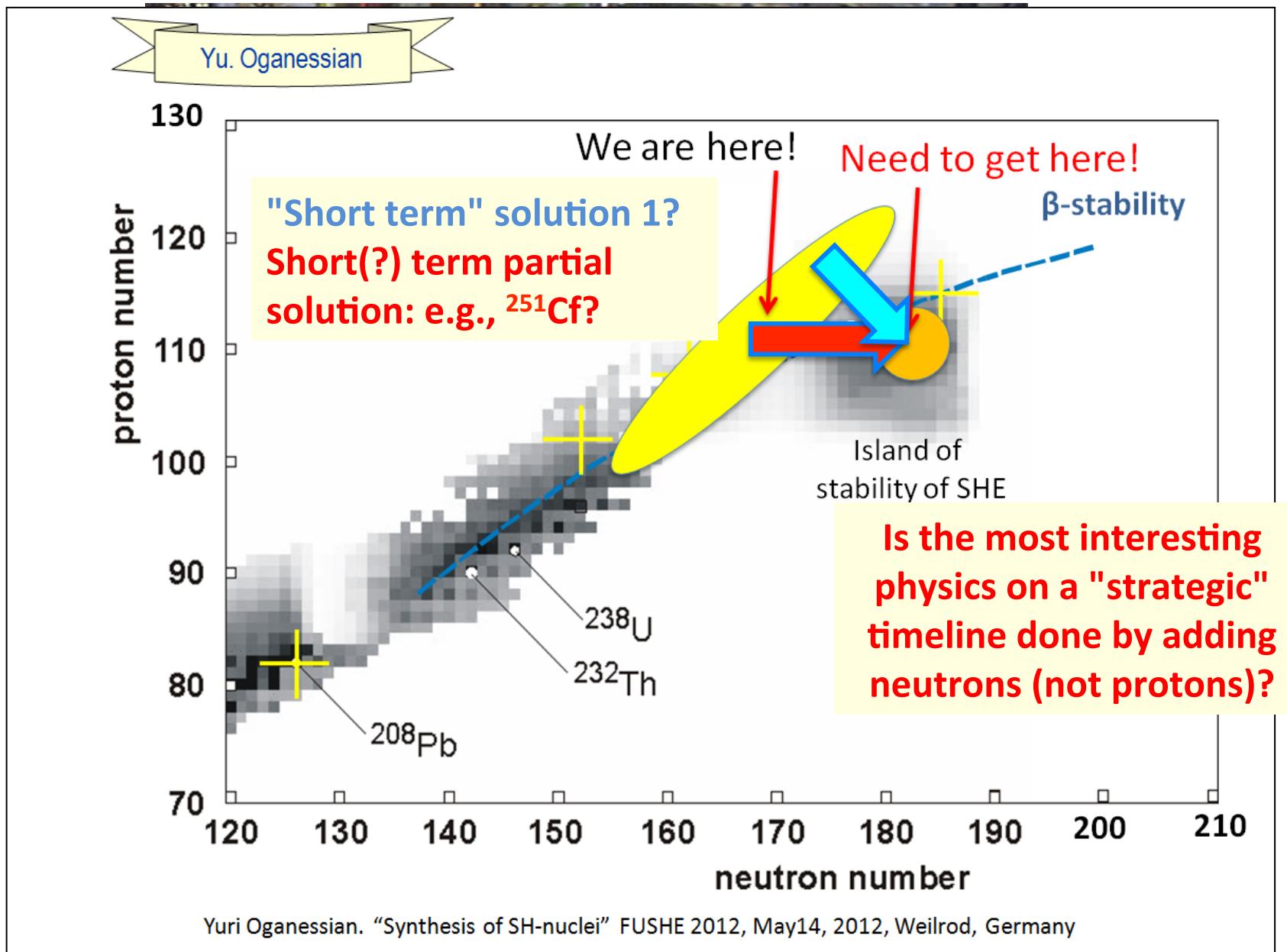
ITE Warsaw (PL)

L.G. Sarmiento

M. Wegrzecki

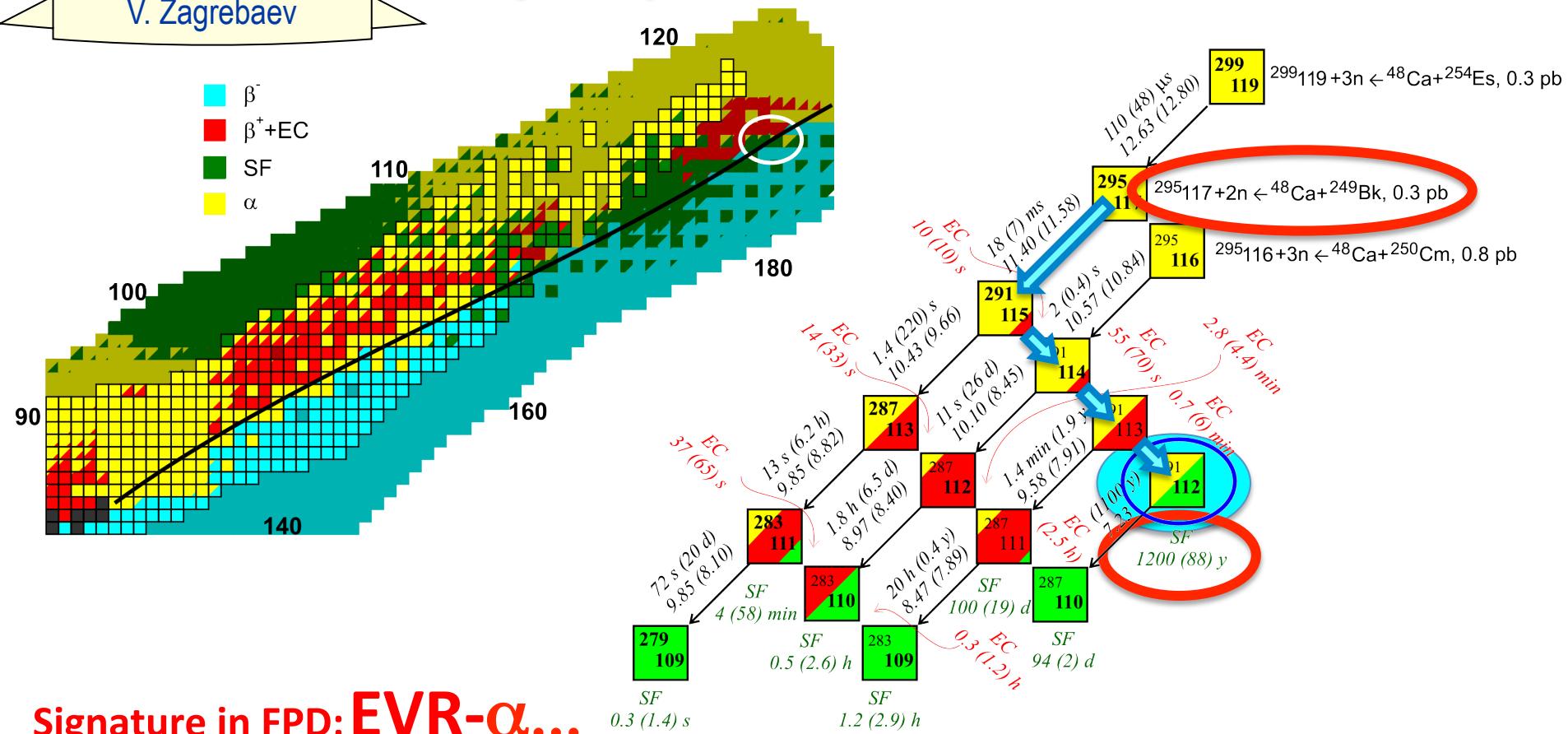


(How) can we produce weighable quantities



Narrow pathway to the island of stability just by fusion reactions !

V. Zagrebaev



Signature in FPD: EVR- α ...

(unobserved

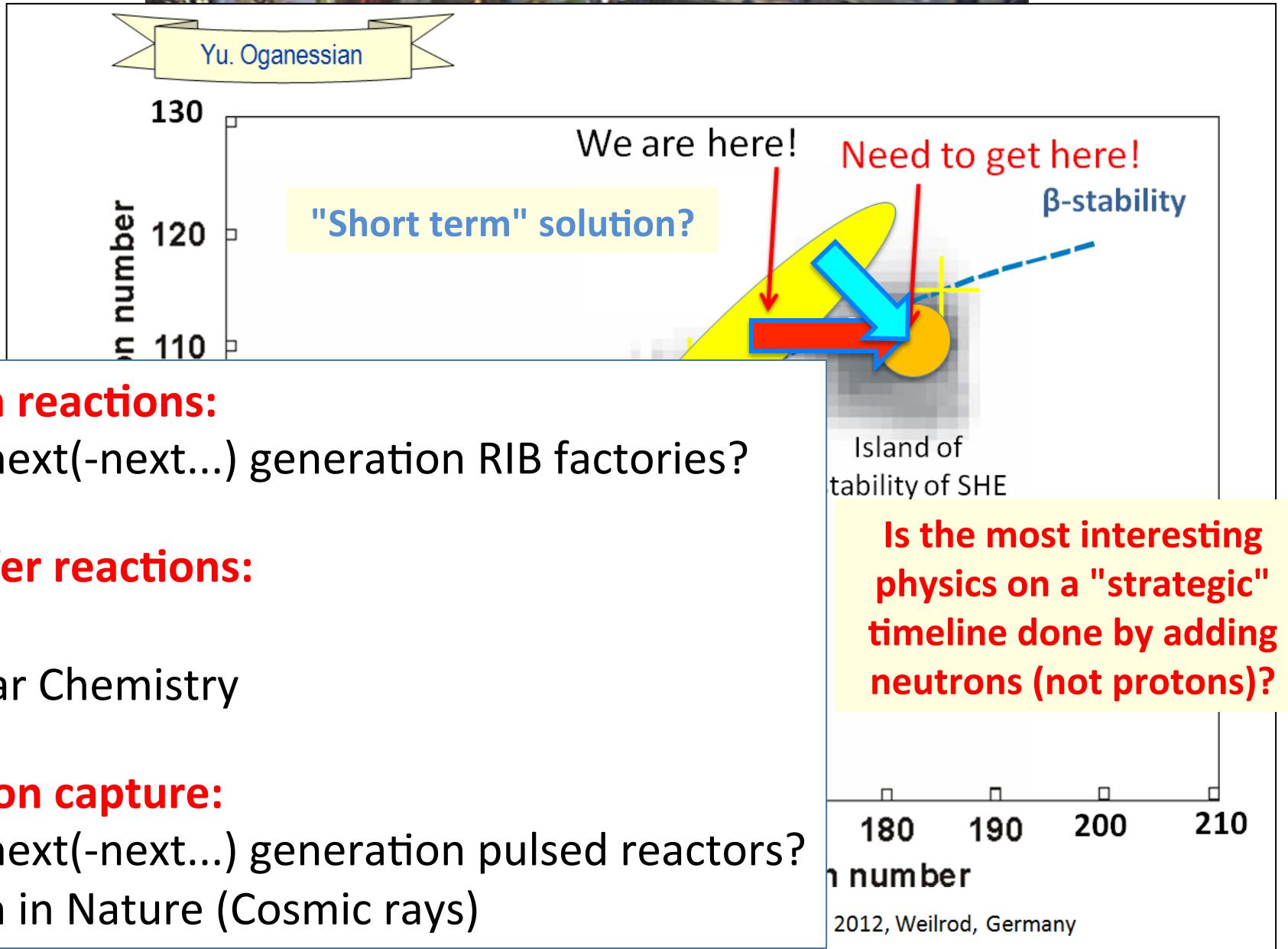
....EC-EC-EC...)

...SHE exists for 1200 a, but nobody knows!!

Need 1: detection system for EC

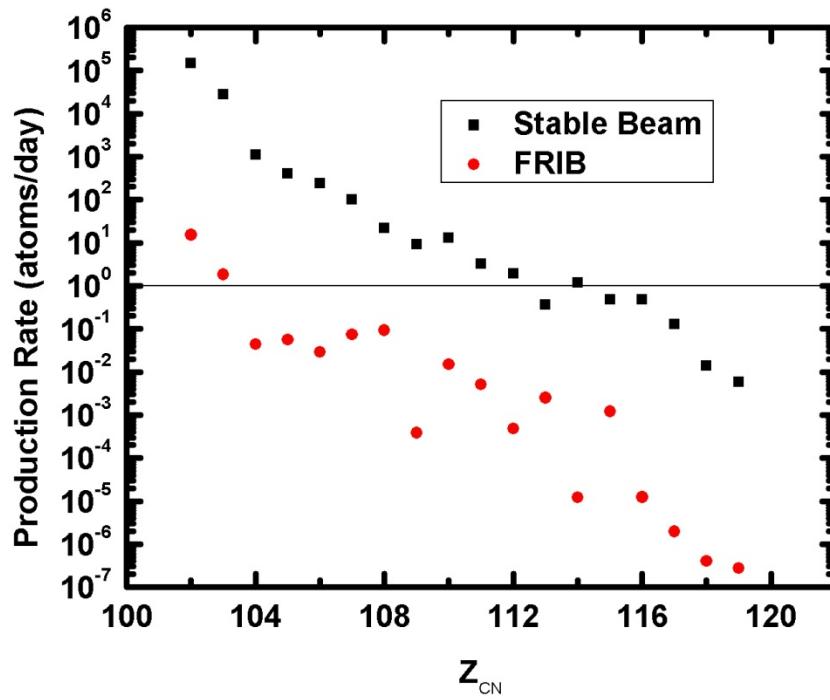
Need 2: non-destructive 1 atom detection / manipulation method!

(How) can we produce weighable quantities

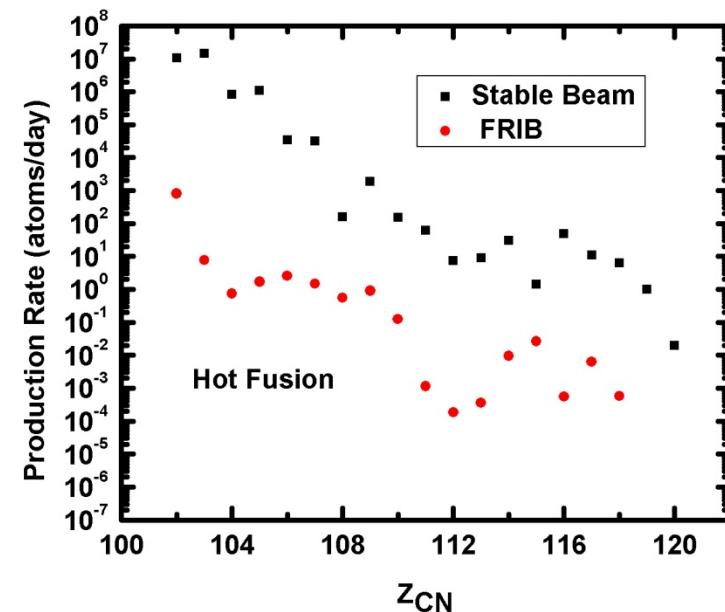


Cold fusion

W. Loveland



New elements from RIB facilities
(LOL)



Detection

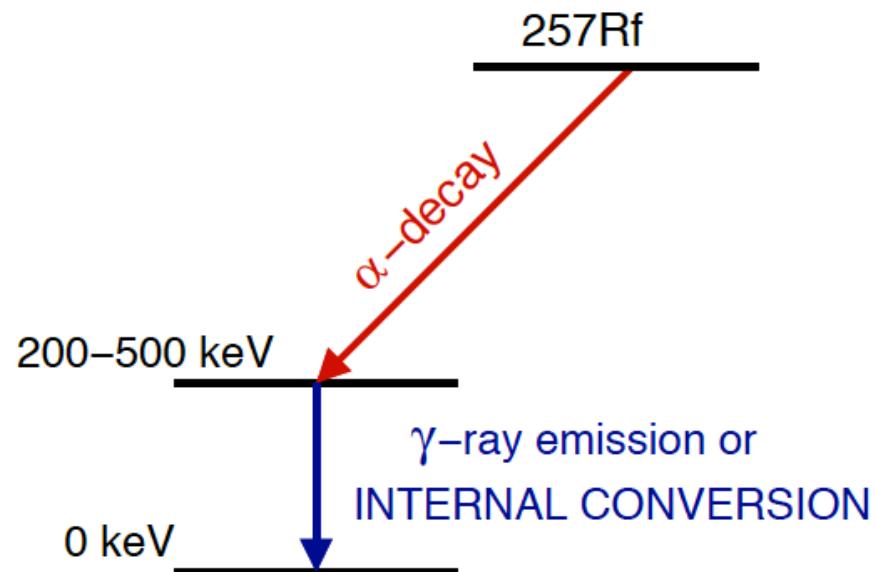
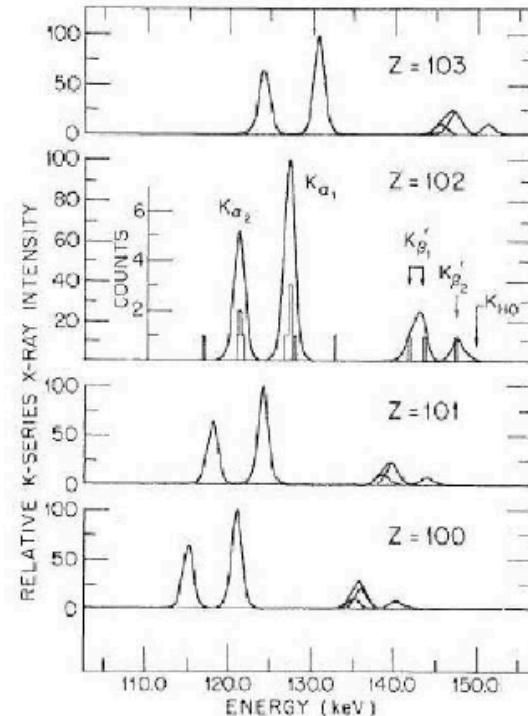
- Shortest halflives:
 - Survival through separator?
 - Disentangling of fast decays?
- Longest halflives:
 - Detection by decay chain not optimal
- Identification through:
 - Chemical ID
 - Mass measurement
 - X-ray fingerprinting
 - cross bombardment
 - alpha chains

X-ray Fingerprinting of Superheavy Nuclei

Courtesy: Spokesperson D. Rudolph, Lund Uni, Sweden

in the spirit of

R. Bemis *et al.*, PRL31, 647 (1973)



15 alpha-photon coincidences
consistent with No ($Z = 102$) K X rays
Identification of Rf ($Z = 104$)

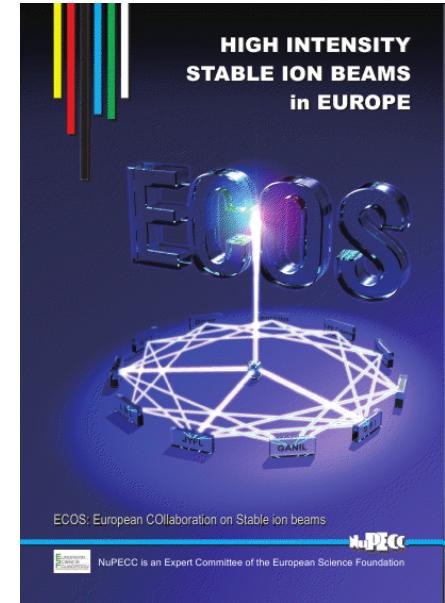


L.-L. Andersson

Rates?

- Factories planned:
 - Dubna
 - GSI
 - Others?

ECOS report recommended dedicated facilities

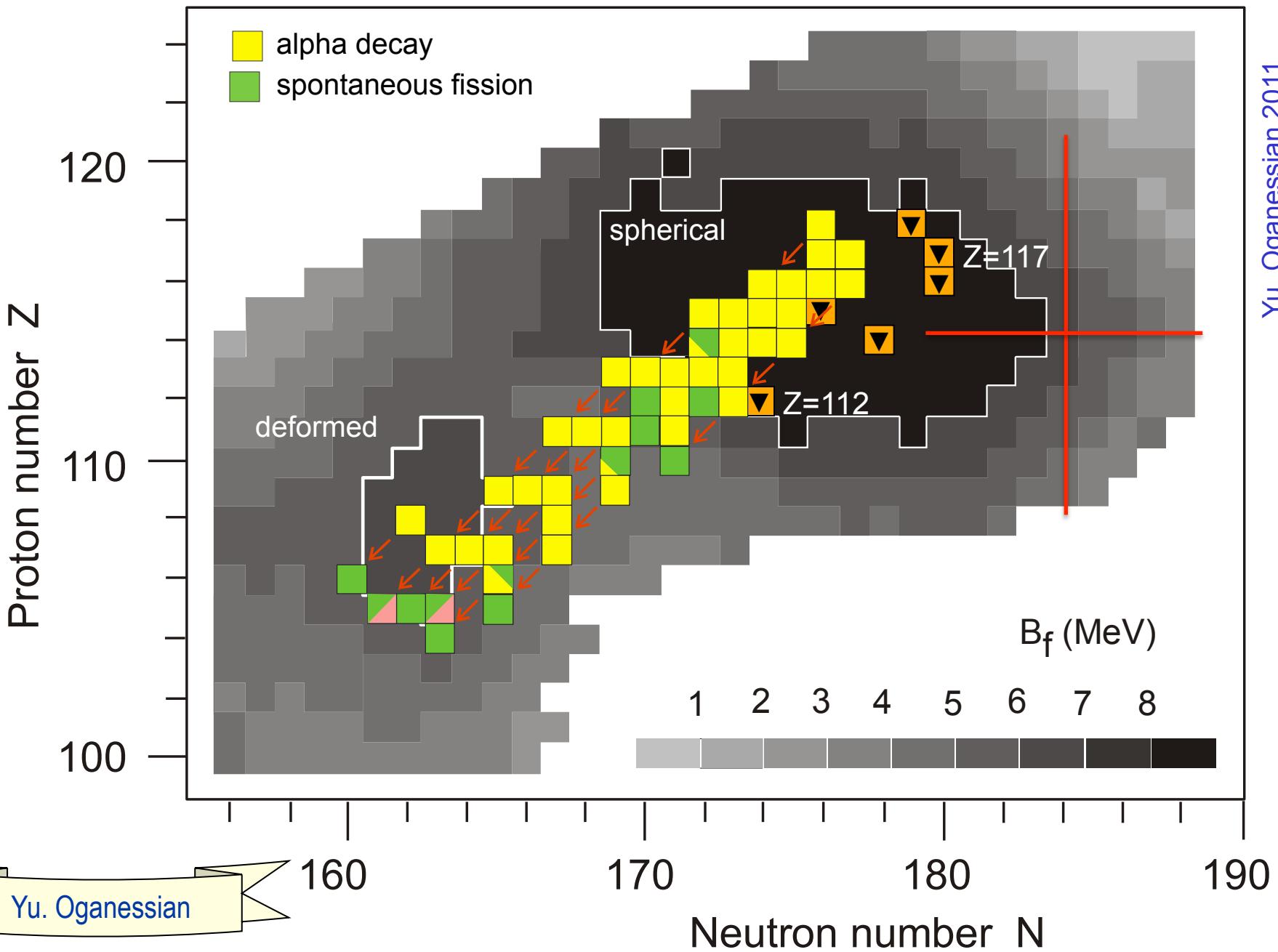


What do we use the extra beamdose for?

- > Push boundaries
- > Precision studies

Spherical Shells

- Where is the island of stability?
 - Where is the shoreline?
 - Where is the peak?
- Experimental body of data still very neutron deficient



Yu. Oganessian

Fission

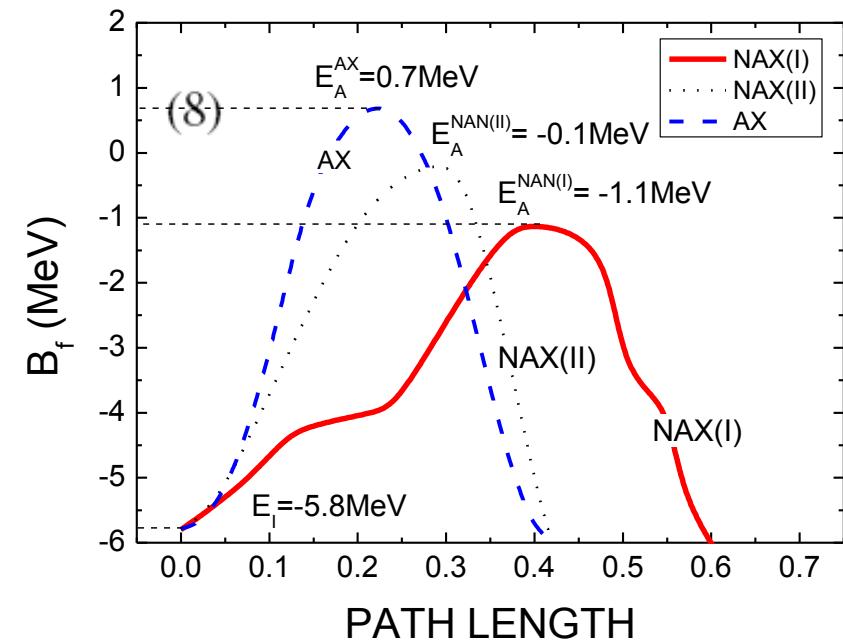
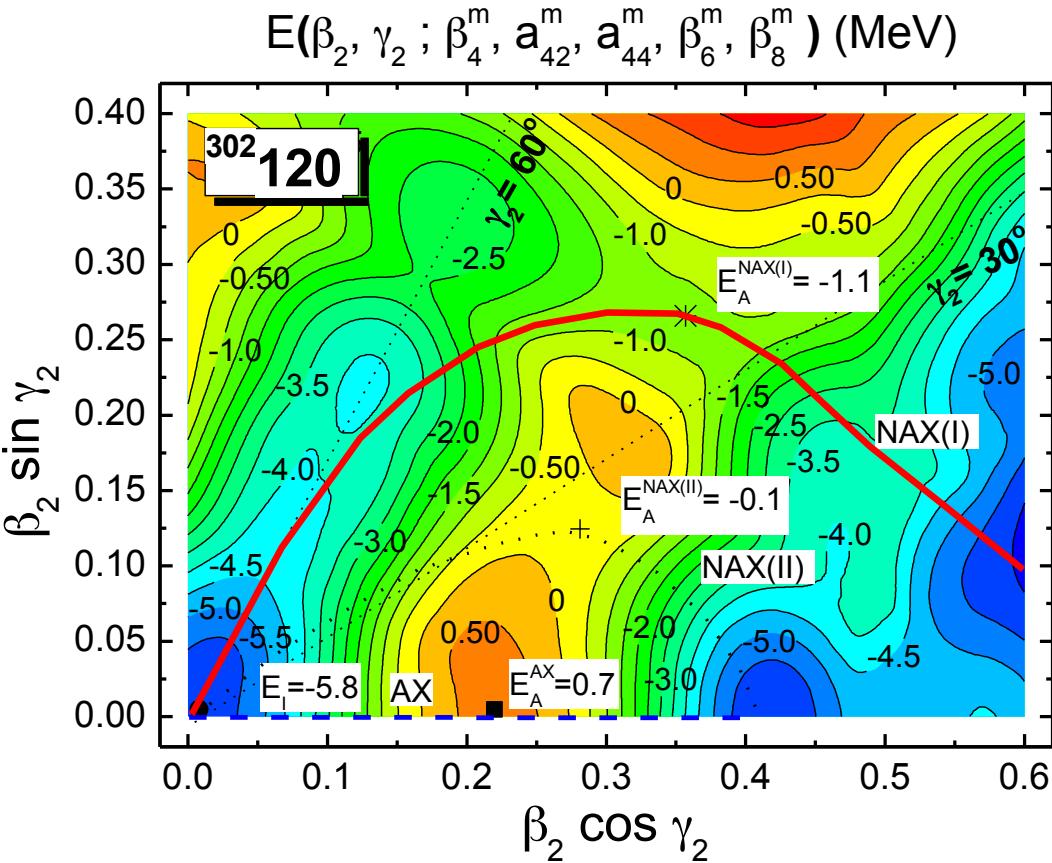
- Fission barriers are key to the problem.
- Experiment: Half-life and height
- BUT: underlying structure is highly complex!

The potential energy is calculated in the following grid points:

$$\beta_2 \cos \gamma_2 = 0(0.05)0.65,$$

$$\beta_2 \sin \gamma_2 = 0(0.05)0.40,$$

$$\beta_4 = -0.20(0.05)0.20.$$



What is the nuclear structure in SHE?

- What is the single particle structure of SHE?
- What are the collective properties?
- What role do isomeric states play in SHE?

Nuclear Structure

- Provide non-integral data
- Looks at lower mass systems:

“Push the Fermi level towards the major shell gaps”

- Rod Clark

In-beam spectroscopy

- Rotational bands
- Configurations of excited states
- Potential to provide the most detailed and complete picture of nuclear structure

How to make progress with in-beam experiments

Need:

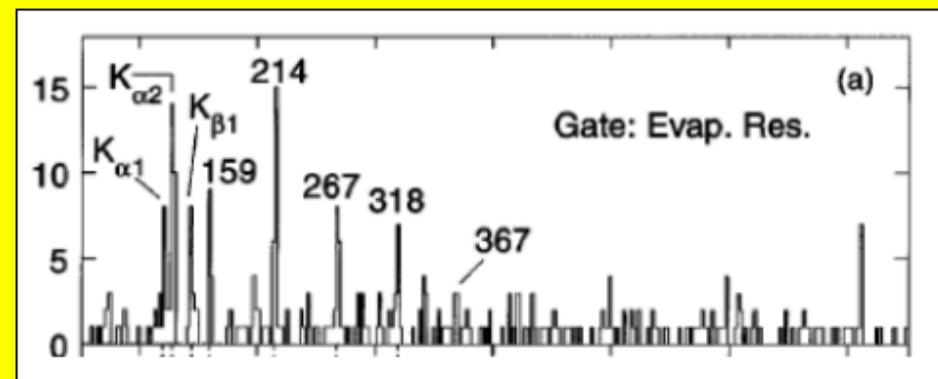
Better separators (but can only gain a factor of 2 or so in transmission)

Better arrays (ultimate gain factor of 5 in singles)

AGATA, GAMMASPHERE,
GRETINA, EXOGAM

Higher beam intensities (digital electronics a must)

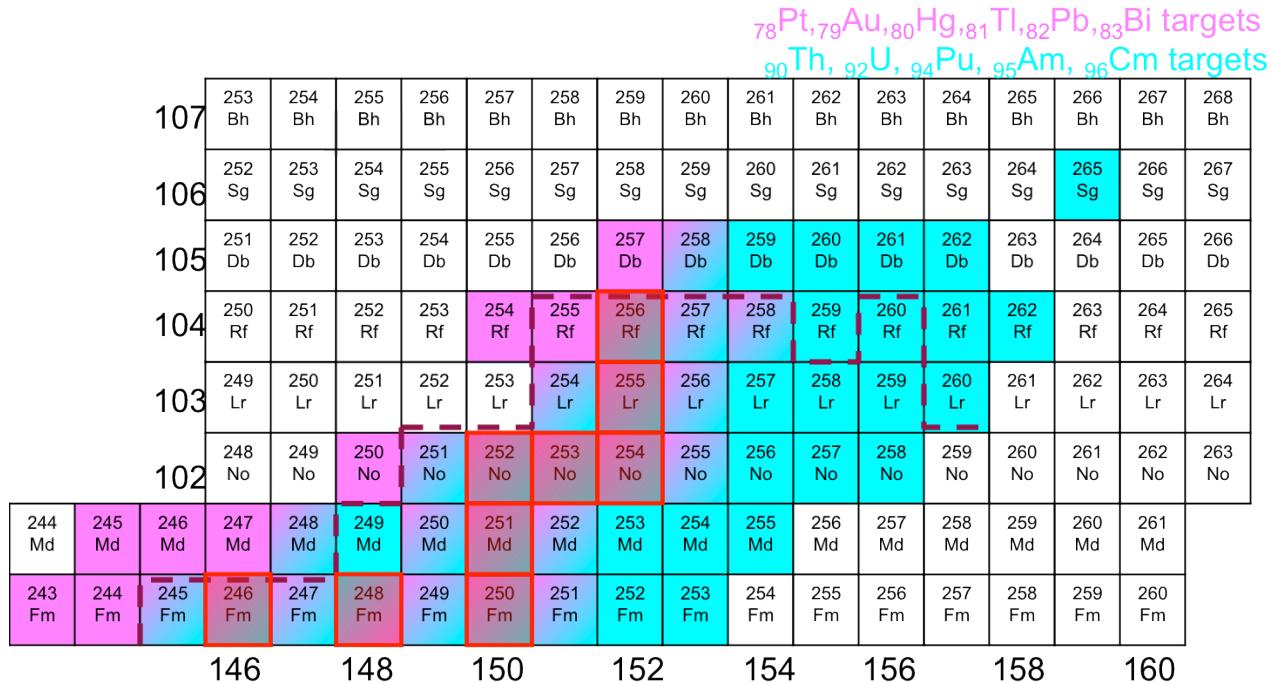
One can maybe reach cross sections of ~ 1 nb



P. Reiter *et al.*, PRL 82, 509 (1999)

Issues

- availability of long-lived (trans)actinide targets
- maximum allowed activity at various facilities and other security issues
- count rates in the arrays due to the activity of the target
- What statistics is required to perform “meaningful” & unambiguous spectroscopy ?



Decay Spectroscopy

- Provides the link between Q-values and masses
- Traces single particle structure
- Can be used with highest beams and smallest cross sections
- Can be used with other methods (chemical, traps)

Physics Motivation for Decay and Nuclear Structure Investigations in the Region of SHE

Why Decay and Nuclear Structure investigations ?

F.P. Hessberger

$B_f(LQ) \rightarrow 0$ for $Z \approx 106$

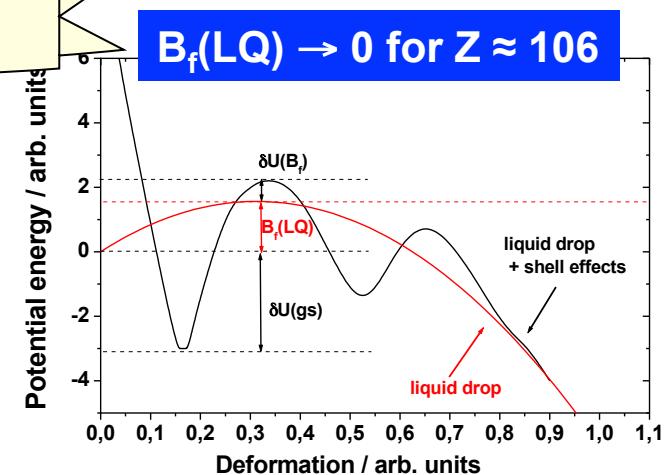
→ Atomic nucleus is quantum mechanical ensemble of nucleons (protons, neutrons)

→ Properties determined by ‚fundamental‘ interactions

- nucleon – nucleon interaction
- Coulomb interaction
- spin – orbit interaction
-

→ Understanding decay properties and nuclear structure – Understanding ‚fundamental‘ interactions

→ Superheavy nuclei (SHE) = ensembles of ‚extremely‘ large numbers of protons and neutrons



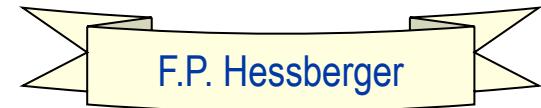
→ Nuclear stability limited by fission
- Fission barriers in SHE determined by shell structure; B_{sf} depends on single particle levels

→ mass excess (-> determines Q-values for α- and β- decay)

→ Understanding ‚basic‘ decay properties essential for understanding limits of stability

→ Understanding nuclear structure is essential for understanding properties and stability of SHE

Decay spectroscopy



→ α - and α - γ – spectroscopy

Q-values, shell crossings;

Information on low lying Nilsson levels from energy and HF for α -decay as well as from energy, intensity and multipolarity of γ -rays.

→ systematic trends in odd mass – even Z nuclei along isotone line

→ systematic trends in odd mass – odd Z nuclei along the isotope line

Comparison with theory

→ Spontaneous fission

Fission barriers, hindrance factors, TKE

→ EC – decay

Often population of higher lying levels than by α -decay with notable intensities; similar information,

implantation method requires CE – K X-ray – γ - coincidences for identification

→ K – Isomers

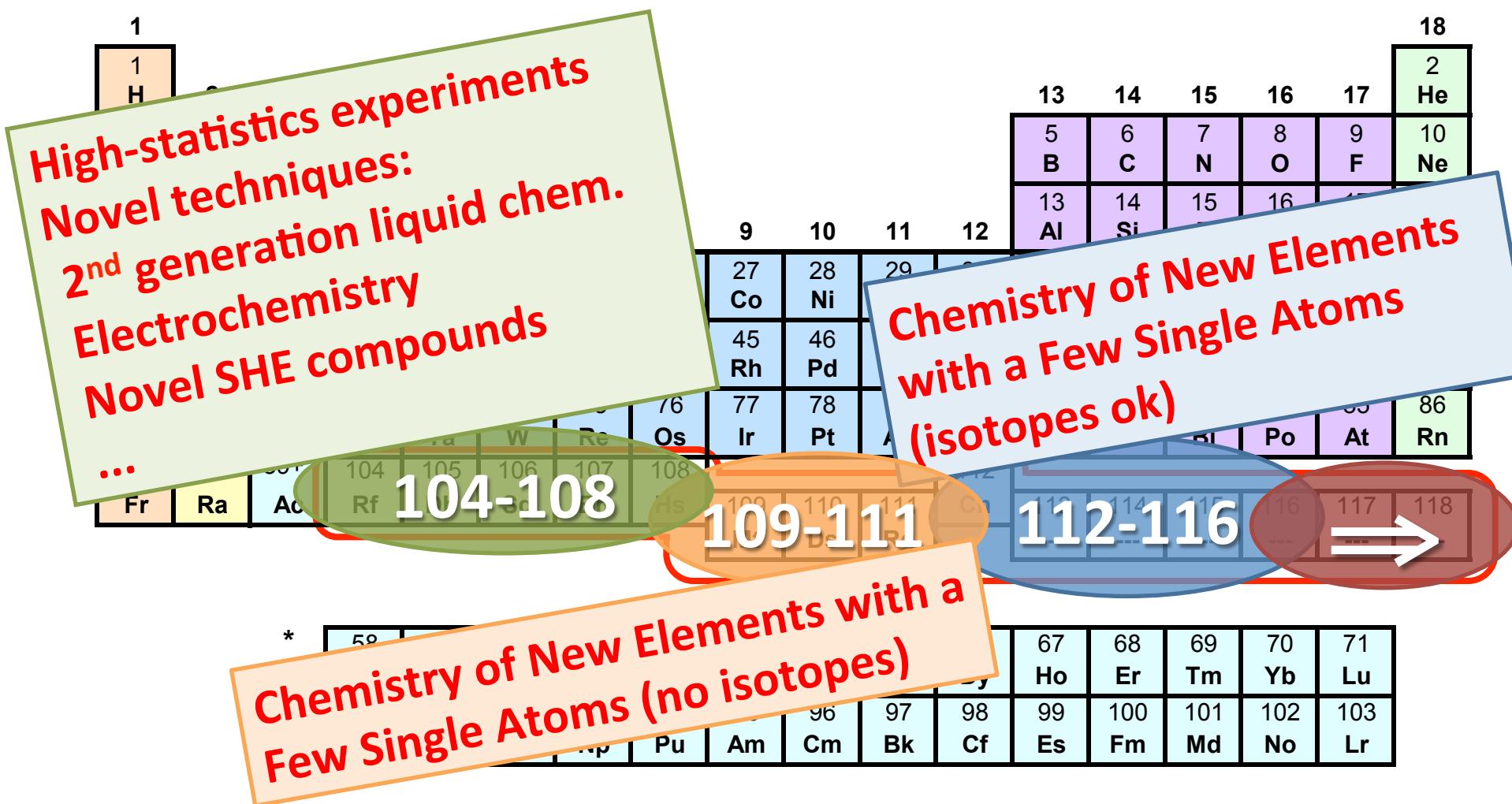
Multi – quasiparticle (2q-proton, 2q-neutron,) configurations, typically at $E^* > 1$ MeV; information on lower lying qp-states, Nilsson levels, level ordering, vibrational states and bands built up on them

CE – γ coincidences required

What are the chemical and atomic properties of the heaviest element?

- With their position in the Periodic Table being fixed by Z , do the SHE's chemical properties conform to those of the lighter homologs?
- What is the influence of relativistic effects?
- How well are these – and their consequences – understood

Superheavy Element Chemistry



For a real understanding of SHE chemistry:
relevant theoretical work

104-108

109-111

112-114

115+

$T_{1/2}$

direct

10 s – 1 min

<< 1s

few s

0.1 s

indirect

1 day

1 s

up to 30 s

0.2 s

Cross section

10s nb-pb

"small"

few pb

1 pb

Interest of SHE chemistry community

High

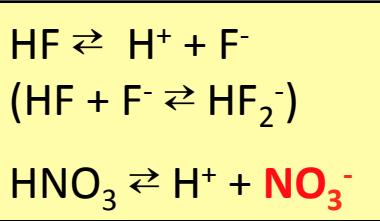
Negligible

Very high

to come

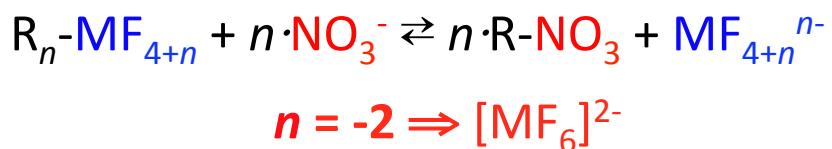
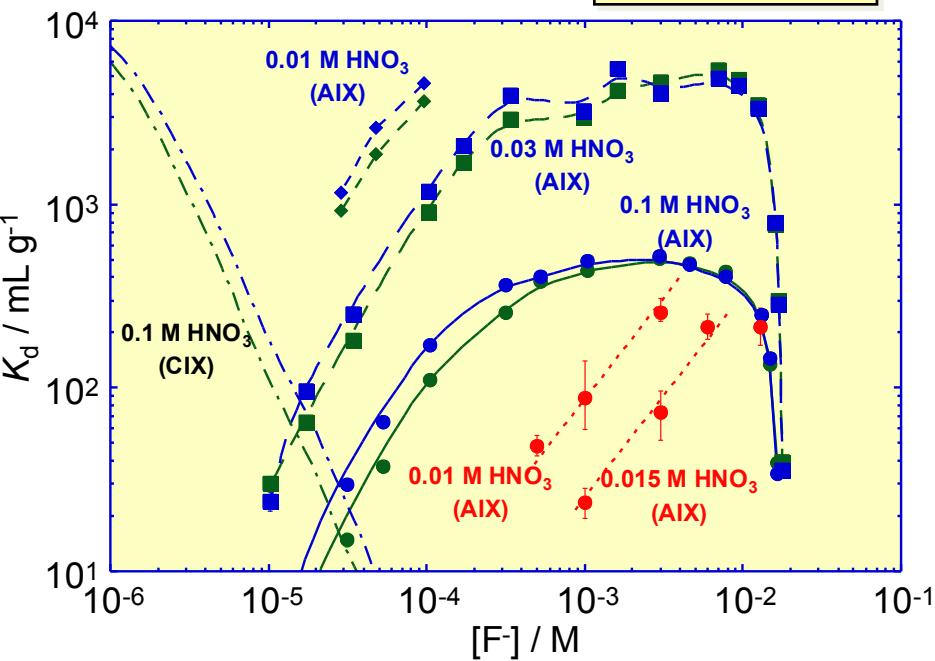
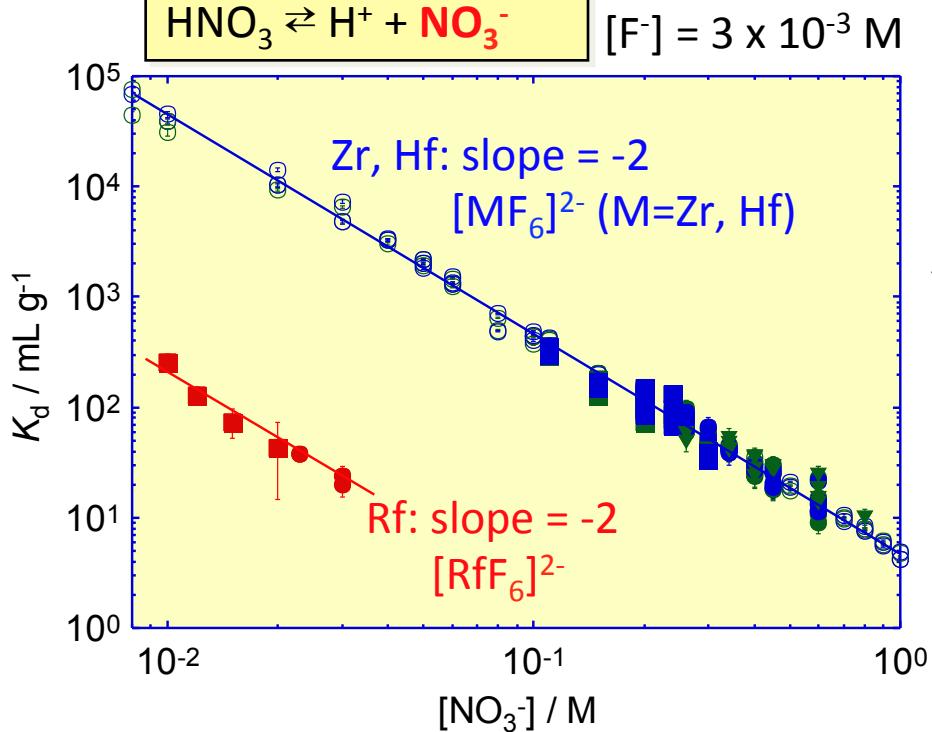
Many techniques

Anion-exchange behavior of Rf in HF/HNO₃



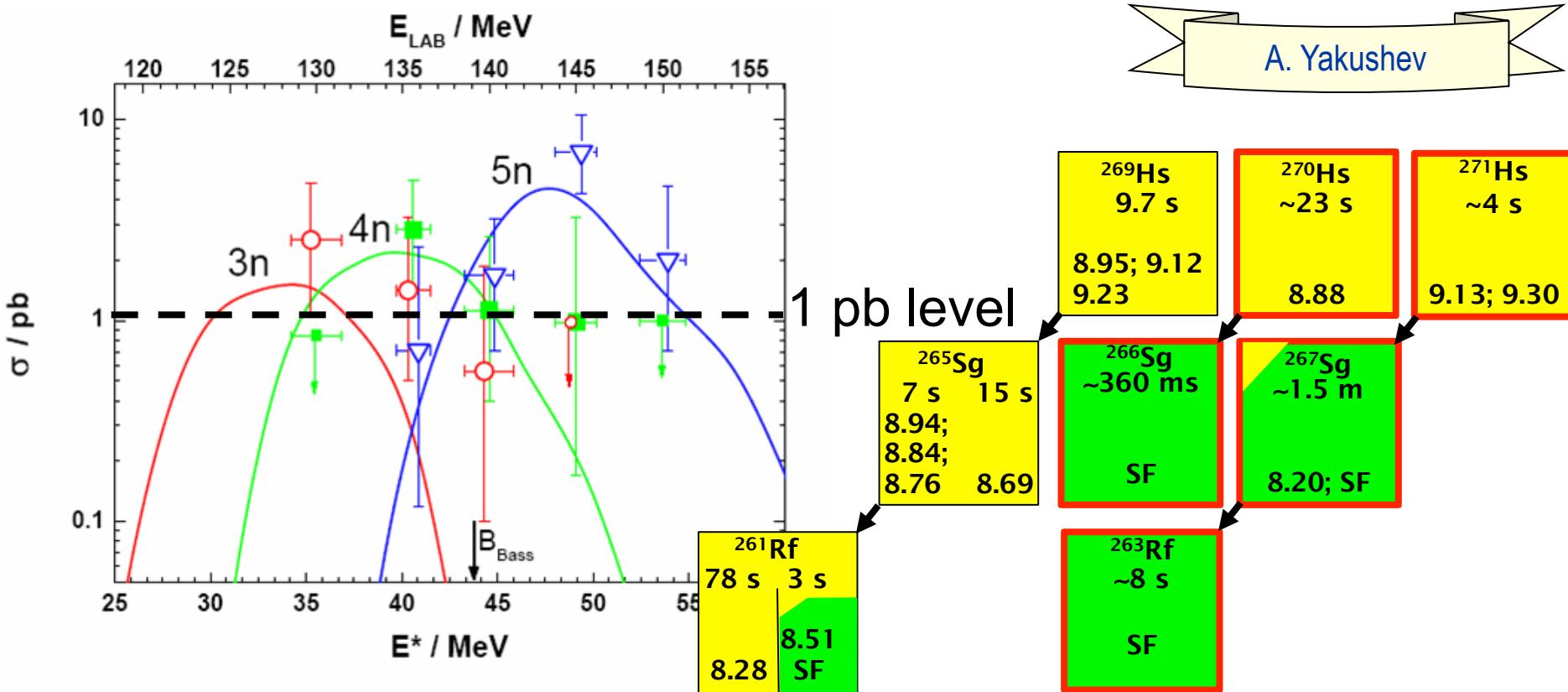
Y. Nagame

Rf (on-line)
 Zr (off-line)
 Hf (off-line)



Formation of $[\text{MF}_6]^{2-}$: Zr \approx Hf $>$ Rf

Chemistry experiment @ high sensitivity



25 decay chains from $^{269,270,271}\text{Hs}$ were detected within 5 weeks
Excitation function was measured @ 5 beam energies

Chemistry of new Elements (Z=109-111)

This page intentionally left blank

PSI-FLNR results Cn+114 (2006-2008)

Pb
~1000°C

(<<<)

Hg
~ 200°C

<<

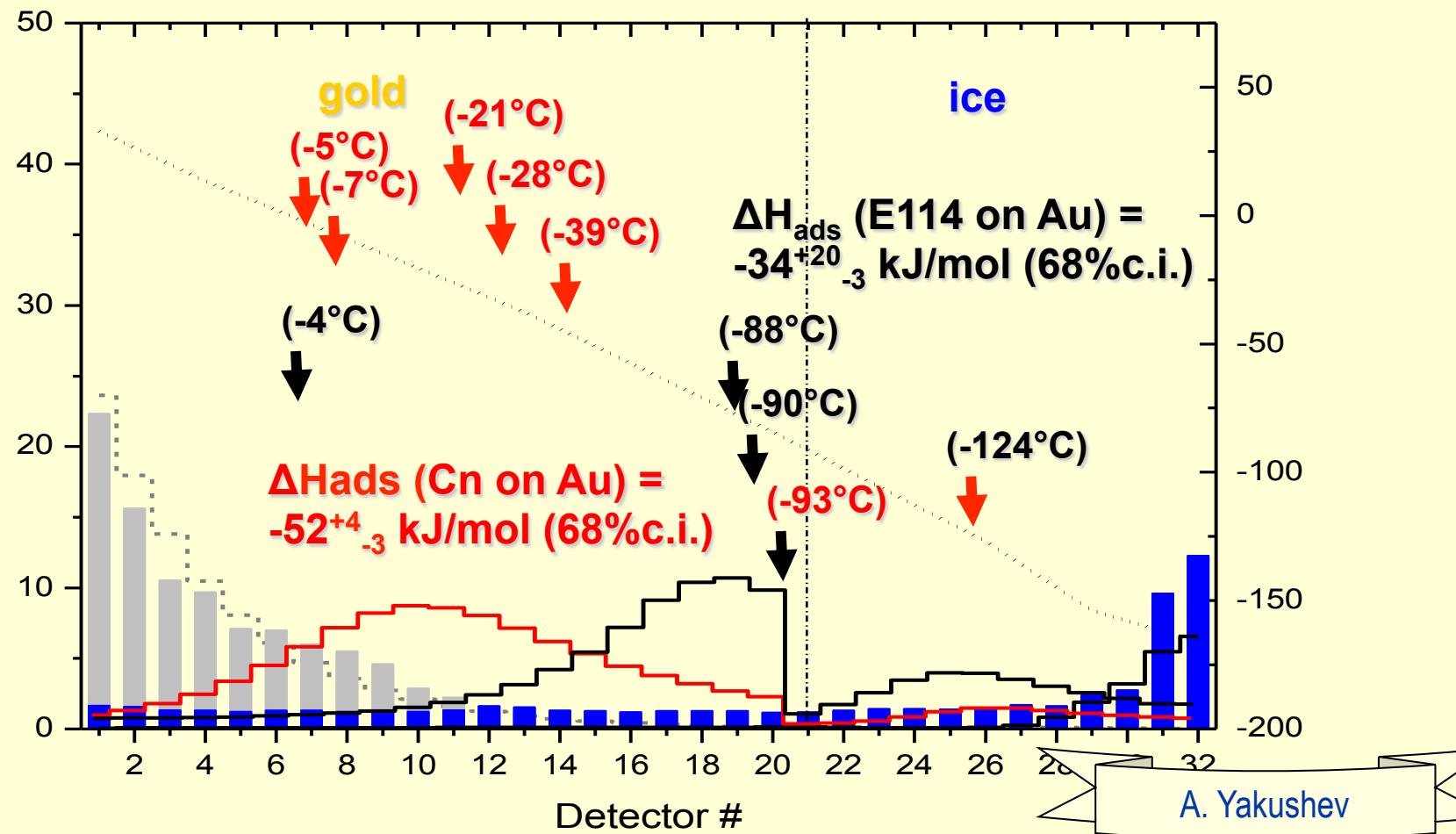
Cn
-25°C

<

114
-90°C

<

Rn
-170°C



Chemistry of new Elements (Z=112+)

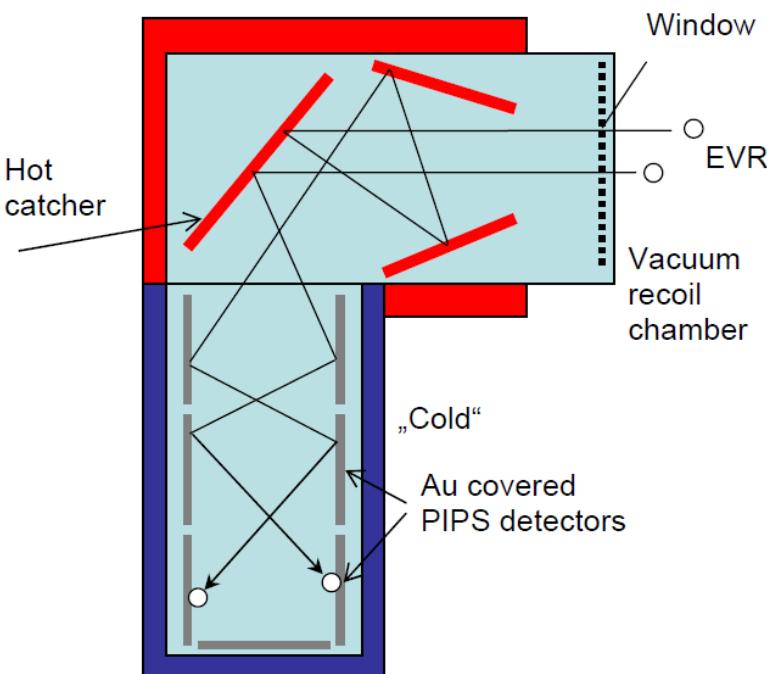
$T_{1/2} \geq 0.5$ s

$T_{1/2}$: 50–500 ms

Uuo 118	118-294 0.89 ms α 11.65; (sf≤50%)
Uus 117	117-292 14 ms α 11.03 117-294 78 ms α 11.40
Uuh 116	116-290 7.1 ms α 10.80 116-291 18 ms α 10.74 116-292 18 ms α 10.66 116-293 61 ms α 10.54
Uup 115	115-287 32 ms α 10.59 115-288 87 ms α 10.46 115-289 0.22 s α 10.21 115-290 16 ms α 9.95
Uuq 114	114-286 0.10 s α 10.19; sf(50%) 114-287 0.48 s α 10.02 114-288 0.69 s α 9.94 114-289 2.1 s α 9.80
Uut 113	113-282 73 ms α 10.63 113-283 100 ms α 10.12 113-284 0.48 s Cn 282 0.82 ms sf 113-285 5.5 s α 9.74; 9.48 Cn 283 3.8 s α 9.52; sf(40%) 113-286 20 s α 9.62 Cn 284 99 ms sf 113-287 29 s α 9.15 Cn 285 29 s α 9.15
Cn 112	

Z=112 – Z=114:
COMPACT / COLD
approach ✓

Z≥115: Vacuum chemistry?



SHE Chemistry

What are the most important experiments possible with present facilities?

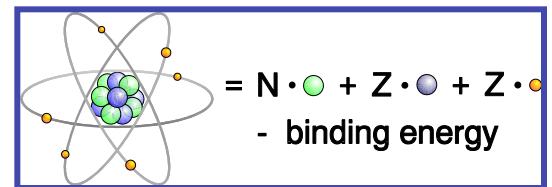
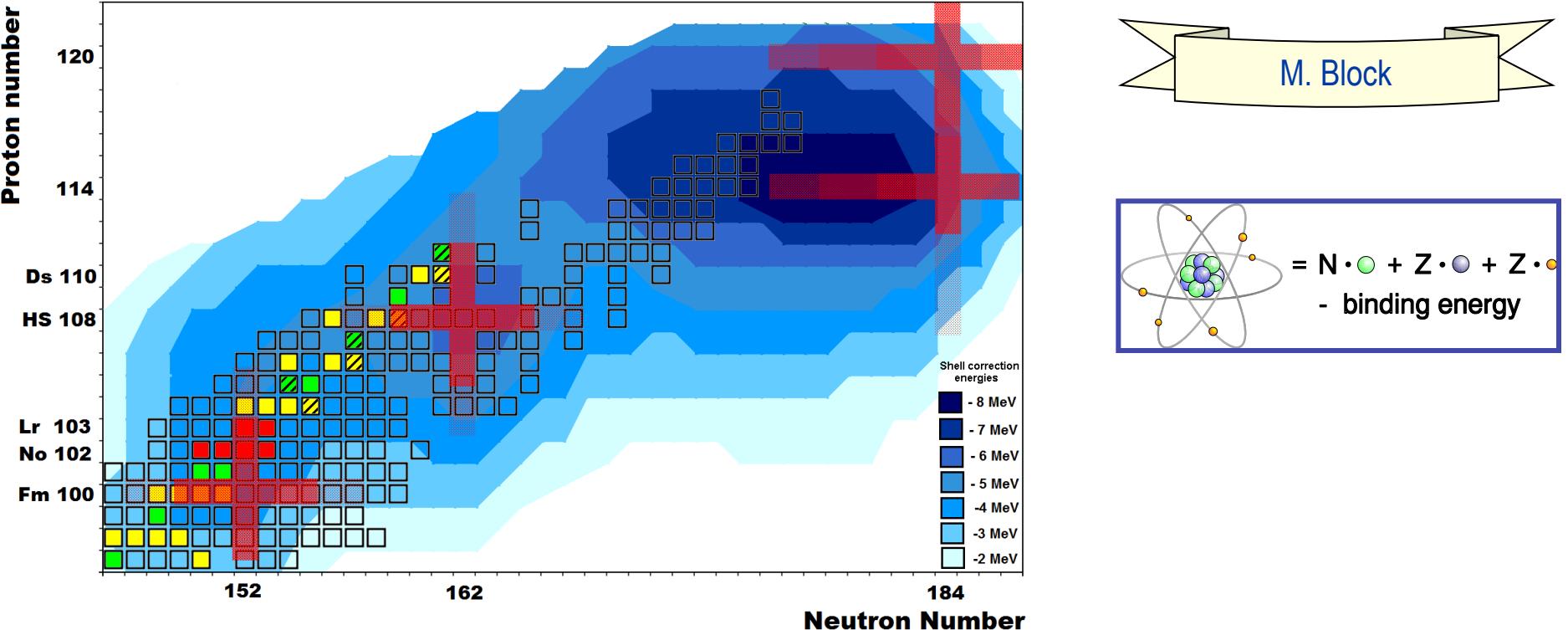
(-> depends on your facility...)

Which exciting experiments will be enabled with planned accelerator and equipment upgrades?

- > experiments with "unknown" elements
- > new techniques (el.chem; preseparation; ...)
- > high-statistics exps. beyond Db
- > new SHE compound classes
- > access to so far inaccessible observables (mass,...)
- > chemical separators for nuclear physics

In which directions should future developments proceed to maximize scientific opportunities?

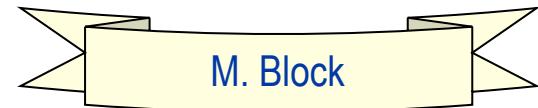
Importance of Masses for $Z > 100$



- masses provide absolute nuclear binding energies
- masses allow studies of the shell structure evolution
- high-precision mass measurements provide anchor points to fix decay chains
- benchmark nuclear models

Nuclear properties from laser spectroscopy

Hyperfine structure interaction



1. Magnetic Dipole HFS

$$A = \frac{\mu \langle H_e(0) \rangle}{I J} \Rightarrow \text{Nuclear magnetic moment } \mu$$

2. Electric Quadrupole HFS

$$B = e Q_s \langle \varphi_{jj}(0) \rangle \Rightarrow \text{Spectroscopic quadrupole moment } Q_s$$

3. Coupling

$$\frac{1}{2} J + \frac{1}{2} I = \frac{1}{2} F \Rightarrow \text{Nuclear spin } I$$

Isotope shift

$\langle r^2 \rangle^{A,A'}$ change of mean square charge radius

Challenges (I)

- Create elements beyond E118.
- Fill the gap between the isotopes produced in cold and hot fusion.
- Synthesize more neutron rich isotopes.
- What role can RIBs play?
- Can we shift the reaction paradigm again?

Challenges (II)

Structure:

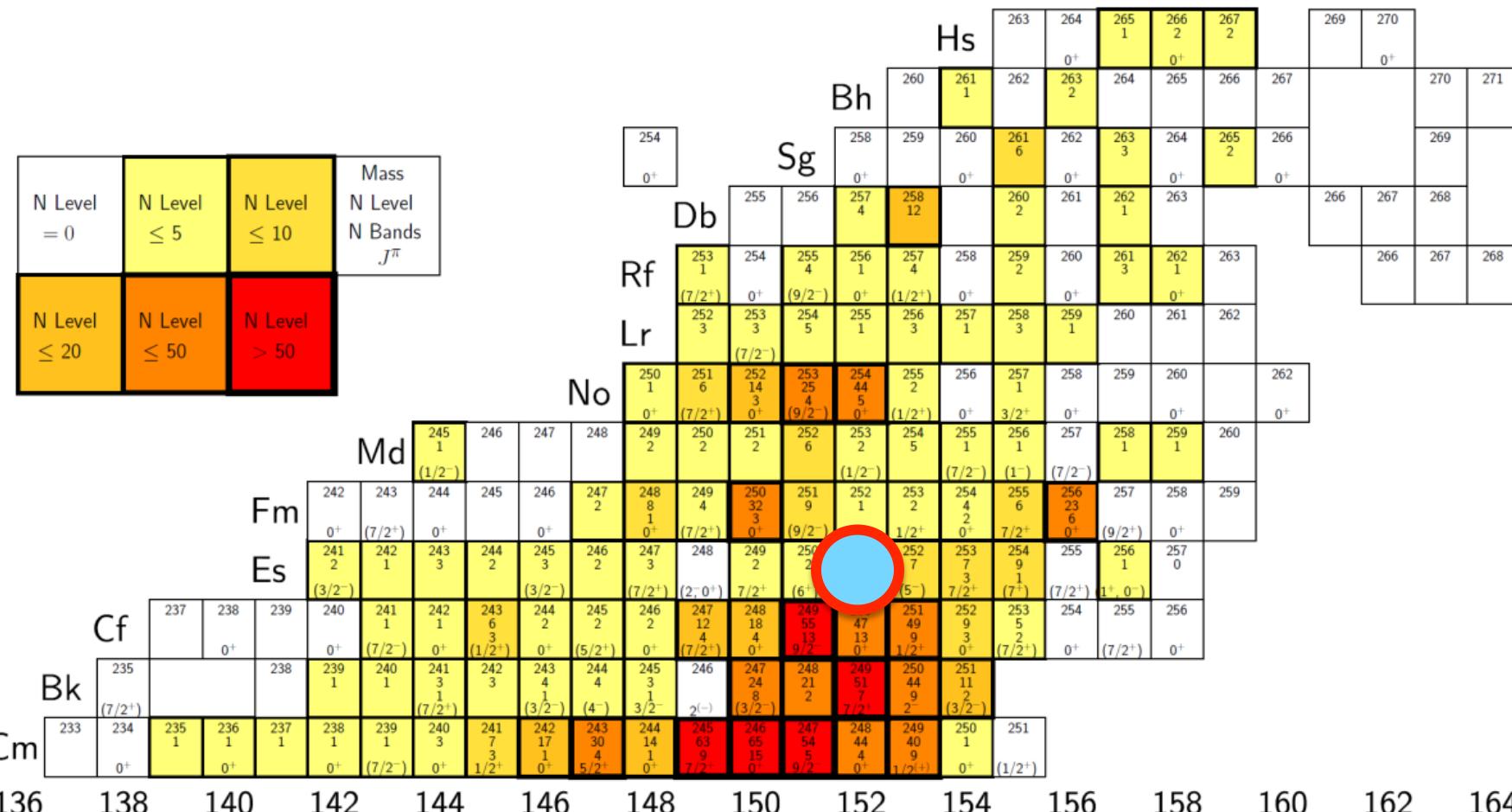
- What are the experimental uncertainties? What data can reliably be used by theorists?
- What new experimental data is needed? Is there guidance from theory in that respect?
- Can we use alternative (optical) methods to extract reliable information on structure?

Urgently needed:(Firm) determination of key observables

M. Leino

Spin-parity: ^{253}Es $I^\pi = 7/2^+$ from optical spectroscopy, magnetic moment
 How to proceed towards higher Z?

Proton Number



Neutron Number

P. v. Duppen

Challenges (III)

- Chemistry
 - Increase production rate for long-lived isotopes!
 - Make new elements!
 - New observables?
 - Direct display of relativistic effects: best done where?
 - Direct chemical speciation
 - What new experimental data is needed? Is there guidance from theory in that respect?

Training and Education

- Remember Walter Loveland's question:

Where does the next generation of SHE scientists come from?